

# Homework 4.

4-2.  $C_6H_{12}O_6$ :  $12 \times 6 + 1 \times 12 + 16 \times 6 = 180 \text{ g/mol}$

$3.5 \text{ g/L}$ ,  $21 \text{ m}^3/\text{d}$ . Biological contactor.

$C_5H_7O_2N$ :  $12 \times 5 + 1 \times 7 + 16 \times 2 + 14 = 113 \text{ g/mol}$

$Y = 0.375 \text{ g biomass / g sugar}$

$A = 2000 \text{ m}^2$

$B_A = 8-12 \text{ g BOD}_5/\text{m}^2/\text{d}$



$\therefore \frac{5.04 \times 32}{1.674 \times 180} = 0.536 \text{ g } O_2/\text{g sugar}$

$\therefore BOD = 0.536 \times 3.5 = 1.875 \text{ g } O_2/\text{L}$

$A_{BC} = \frac{Q_0 \cdot C_{O, BOD}}{B_A} = \frac{21 \times 1000 \times 1.875}{B_A} = \frac{39375 \text{ g/d}}{B_A}$

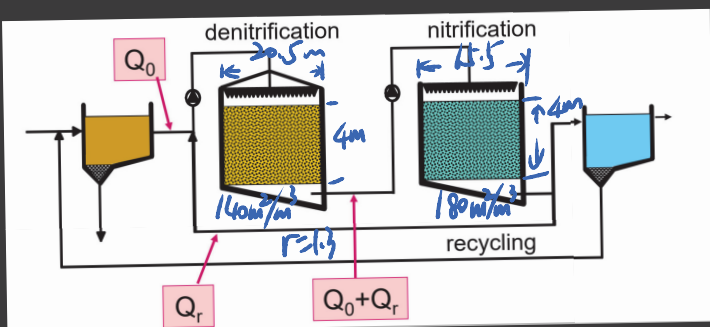
$= 3281.25 \text{ m}^2 \sim 4 \times 21.875 \text{ m}^2$

each is  $2000 \text{ m}^2$ . so require 2.

## 4-3 bio trickling filter, + pre-nitrification.

$Q_0 = 5000 \text{ m}^3/\text{d}$   $C_{O, BOD} = 170 \text{ g } O_2/\text{m}^3$   $N_{TKN, inf} = 45 \text{ g } N/\text{m}^3$

$N_{NO_3, eff} = 15 \text{ g } N/\text{m}^3$   $N_{NH_4, eff} = 2 \text{ g } N/\text{m}^3$



$\phi'_{De} = 20.5 \text{ m}$ ,  $h_{De} = 4 \text{ m}$ ,  $a_{s, De} = 140 \text{ m}^2/\text{m}^3$

$\phi'_{N} = 15.5 \text{ m}$ ,  $h_{N} = 4 \text{ m}$ ,  $a_{s, N} = 180 \text{ m}^2/\text{m}^3$

$$C_{den} = C_{O,TKH} + C_{O,N_2} - 0.045 \cdot C_{O,BOD_5} - C_{e,TKH} - C_{e,N_2}$$

$$= 45 \times 10\% + 0 - 0.045 \times 170 \times 75\% - 2 - 15$$

$$= 17.7625 \text{ g}_N/\text{m}^3$$

$$r_{den} = \frac{C_{den}}{C_{O,BOD_5}} = \frac{17.7625}{170 \times 75\%} = 0.139 \text{ g}_N/\text{g}_{BOD_5}$$

$$B_{A,BOD_5} = e^{(0.182 - r_{den})/200} = e^{1.075} = 2.93 \text{ g}_{BOD_5}/\text{m}^2/\text{d}$$

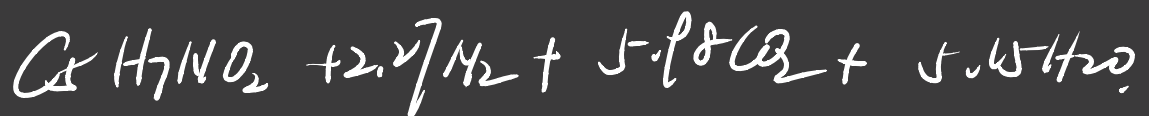
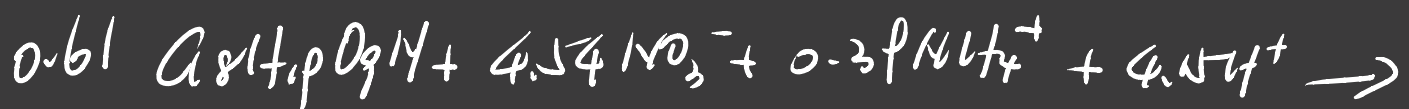
$$A_{den} = \frac{Q_0 \cdot C_{O,BOD_5}}{B_{A,BOD_5}} = \frac{5000 \times 170 \times 75\%}{2.93} = 217577 \text{ m}^2$$

$$V_{TF} = \frac{A_{den}}{Q_{side}} = \frac{217577}{140} = 1554 \text{ m}^3$$

$$V'_{TF} = 2 \times 4 \times \left(\frac{205}{2}\right)^2 \cdot \pi = 2640 \text{ m}^3 \gg 1554 \text{ m}^3$$

$$\phi_{pe} = \sqrt{\frac{V_{TF}}{8\pi}} \times 2 = 15.73 \text{ m}$$

$$\Delta C_{NH_4} = 45 \times 0.9 - 0.045 \times 170 \times 0.75 - 2 = 32.7625 \text{ g}_N/\text{m}^3$$



$$\Delta OIM = \frac{17.7625}{14 \times 4.54} \times 0.61 \times 3.93 = 67 \text{ g}/\text{m}^3$$

$$\therefore \Delta BOD_5 = 67 \times 0.72 = 48.24 \text{ g}_{O_2}/\text{m}^3$$

$$\therefore C'_{O,BOD_5} = 170 \times 0.75 - 48.24 = 79.3 \text{ g}_{BOD_5}/\text{m}^3$$

$$\therefore A_{tot} = A_{BOD_5} + A_{nit} = \frac{Q_0 \cdot C_{O,BOD_5}}{j_{BOD_5}} + \frac{Q_0 \cdot (C_{O,NH_4} - C_{e,NH_4})}{j_{NH_4}}$$

$$j_{\text{BOD5}} = 4-7 \text{ g BOD}_5/\text{m}^2/\text{d} \quad j_{\text{NH}_4} = 0.1 \text{ g NH}_4/\text{m}^2/\text{d}$$

$$\therefore A_{\text{tot}} = \frac{5000 \times 7 \text{ P.3}}{j_{\text{BOD5}}} + \frac{5000 \times 32.76 \text{ P.5}}{0.1}$$

$$= 26865 \text{ m}^2 \text{ n } 28116 \text{ m}^2$$

$$V_{\text{TF}} = \frac{A_{\text{tot}}}{a_s} = 1326 \text{ m}^3 \text{ n } 1562 \text{ m}^3$$

$$V'_{\text{TF}} = 2 \times 4 \times \left(\frac{15.5}{2}\right)^2 \cdot \pi = 1506 \text{ m}^3 \approx V_{\text{TF}}$$

$$r = \frac{C_{\text{dent}} + 1}{C_{\text{e, NH}_3} - 1} = \frac{17.76 \text{ P.5} + 1}{15 - 1} = 1.34 \rightarrow 1.30$$

4-4 AT  $\Rightarrow$  MBBR. 15000 + 10000 capita.

0.35 m<sup>3</sup>/capita. nitrification.

$$Y \rightarrow \boxed{\frac{1000}{1000}} \quad 1100 \text{ m}^3 \times 2. \quad C_{0, \text{TSS}} = 185 \text{ g TSS/m}^3$$

$$Q_0 = 5500 \text{ m}^3/\text{d} \quad C_{0, \text{BOD}} = 160 \text{ g BOD}_5/\text{m}^3 \quad K_{\text{TKN, int}} = 35 \text{ g N/m}^3$$

a).  $\theta_x = 10 \text{ d}$ .  $F_T = 1072^{(10-15)} = 0.706$

$$SP_c = 0.75 + 0.6 \times \frac{185 \times 0.5}{160 \times 0.75} - \frac{0.102 \times 10 \times 0.706}{1 + 0.17 \times 10 \times 0.706}$$

$$= 0.885$$

$$\therefore V_{\text{AT}} = \frac{Q_0 \cdot C_{0, \text{BOD}} \cdot SP_c \cdot \theta_x}{X_{\text{AT}}}$$

$$= \frac{5500 \times 160 \times 0.75 \times 0.885 \times 10}{1000 \times 3} = 1947 \text{ m}^3 < 2200 \text{ m}^3$$

enough.

$$b) V_{\text{BOD}_5, \text{reactor}} = \frac{160 \times 0.75 \times 5500 \times (15+10)}{15 \times 1.3 \times 1000} = 846 \text{ m}^3$$

$$\Delta C_{\text{NH}_4\text{-N}} = 25 \times 0.1 - 0.045 \times 160 \times 0.75 - 2$$

$$= 24.1 \text{ g/m}^3 \quad 27.6 \text{ g/m}^3$$

$$\therefore V_{\text{NH}_4\text{-N}, \text{reactor}} = \frac{24.1 \times 5500 \times (15+10)}{15 \times 0.193 \times 1000}$$

$$= 1145 \text{ m}^3 \quad 1311 \text{ m}^3$$

$$V_{\text{tot}} = V_{\text{BOD}_5, \text{reactor}} + V_{\text{NH}_4\text{-N}, \text{reactor}}$$

$$= 846 + 1145 = 1991 \text{ m}^3 < 2200 \text{ m}^3$$

enough.

$$2156 \text{ m}^3$$